

# **EZ3505 Chloride Analyser**

Method and reagent sheets 01/2023, Edition 1.01

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# 1. Legal information

Manufacturer: AppliTek NV/SA Distributor: Hach Lange GmbH

The translation of the manual is approved by the manufacturer.

# 2. Analytical specifications

Please refer also to the respective technical datasheet at Hach Support Online.

Chloride - All specifications	hloride - All specifications			
Analysis method	Disco	ntinuous measurement by combined ion	-selective electrode wi	th standard addition
Parameter	Chlori	de		
Cycle time		ard measurement cycle time: 5 minutes al dilution: + 5 min.	3	
Limit of detection (LOD)	≤ 10 r	ng/l		
Precision/Repeatability	Better	than 2% full scale range for standard te	est solutions	
Cleaning	Auton	natic; frequency freely programmable		
Calibration	Automatic, 2-point; frequency freely programmable			
Validation	Automatic; frequency freely programmable			
Interferences	Bromide [(Br)-], sulphide [(S)2-], iodide [(I)-], cyanide [(CN)-] ions may interfere. Mercury [(Hg)+] must be absent. Ammonia [NH3] and thiosulphate [(S <sub>2</sub> O <sub>3</sub> )2-] may interfere. Fats, oil, proteins, surfactants and tar.			
Measuring ranges	% of ı	range - Dilution	Low range (mg/L)	High range (mg/L)
	В	25% of standard range	10	250
	C 50% of standard range 25 500			500
	0 standard range 50 1000			1000
	V	internal dispenser dilution (factor 5)	250	5000
	W internal dispenser dilution (factor 10) 500 10000			10000

## 3. Analysis method

#### Summary

The Chloride (Cl<sup>-</sup>) concentration is determined by a standard addition using an ionselective electrode.

### **Analysis steps**

The analysis vessel is drained and rinsed with fresh sample. A specific amount of sample is dosed into the analysis vessel. An ISA [T(otal) I(onic) S(trength) A(djustment) B(uffer)] solution is added to the sample to adjust the pH and to assure the total strength of the sample. The potential is measured using an ion selective electrode. Next a known volume of standard solution is added. The solutions are mixed and a second reading is taken. The analyser calculates the initial ion concentration in the sample.

#### Calibration

The calibration procedure measures a REF1 Blank solution (channel 9, REF1 valve) to adapt the slope by means of a one point calibration.

The calibration is performed in the MAIN method.

#### Remark

The methods cannot be started at the same time.

## 4. Reagents

## **ACAUTION**



Chemical exposure hazard. Obey laboratory safety procedures and wear all of the personal protective equipment appropriate to the chemicals that are handled. Read the safety data sheet from the supplier before bottles are filled or reagents are prepared. For laboratory use only. Make the hazard information known in accordance with the local regulations of the user.

## **A CAUTION**



Chemical exposure hazard. Dispose of chemicals and wastes in accordance with local, regional and national regulations.

## 4.1 Reagent overview and consumption

In the tables below, the products that are needed to prepare the reagents are listed. The product name, the formula, the molecular weight, the CAS No. and the amount needed to prepare 1 litre of the reagents is given. Check the consumption of the reagents (28 days) to adapt the volumes needed.

Product	Consumption	Consumption/28 days A rata 1 analysis/5 min	Recommended containers
Buffer solution	~ 0.75 mL / analysis	~ 6 L	Plastic – 10 L
Standard solution Chloride (dispenser)	~ 0.5 mL / analysis ~ 2 mL / calibration	< 5 L	Plastic – 5 L
Blank solution (REF1)	~ 0.5 L / calibration	1	Plastic – 2.5 L

# 4.2 DI-water overview and consumption

	Rinse water (mL/analysis) Type I	Dilution water (mL/analysis) Type I	Total (mL/analysis)	Consumption/28 days A rata 1 analysis / 5 min
В	N.A.	N.A.	N.A.	N.A.
С	N.A.	N.A.	N.A.	N.A.
0	N.A.	N.A.	N.A.	N.A.
V	50 mL	75 mL	125 mL	504 L
W	50 mL	75 mL	125 mL	504 L

## 4.3 Storage and quality of chemicals

#### **Quality of chemicals**

All chemicals should be of Reagent grade, ACS grade or better (\*). The use of pro analysis chemicals is recommended. Poor quality of the reagents can affect the analyser performance.

(\*) Analytical Reagent (AR), Guaranteed Reagent (GR), UNIVAR, AnalaR, Premium Reagent (PR), ReagentCertified ACS reagent, ACS Plus reagent, puriss p.a. ACS reagent, ReagentPlus®, TraceCERT®, Suprapur®, Ultrapur®, or better are also possible.

#### **Quality of DI-water**

All EZ analysers are tested with standard solutions, reagents and dilution water prepared using type I water or better as defined by ASTM D1193-91.

To achieve the specifications as stated on the data sheet, method and reagents sheet and acceptance test reports, the same water quality (or better) must be used for the preparation of the standard solutions, reagents and dilution water.

Additionally the water used for the preparation of the standard solutions for an EZ analyser must be free of the parameter or any of the interferences for the method of that EZ analyser.

#### Storage of Reagents

While operating the instrument, keep in mind the reagent requirements as stated in the reagent overview, the chapters below and/or in the data sheet of the instrument.

## **ACAUTION**



For longer-term storage: Store the reagents cold; Store the reagents in the dark;

If applicable: Store the reagents in a fridge during operation

## **A** CAUTION



Refresh the reagents after one month (unless stated differently in the chapters below).

Do not mix old reagents with freshly prepared reagents. Remove old reagents from the container before adding freshly prepared reagents.

#### 4.4 Buffer solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Potassium nitrate	KNO <sub>3</sub>	101.102	7757-79-1	202.2 g

#### **Preparation**

Add carefully 202.2 g potassium nitrate (KNO<sub>3</sub>) to 400 mL de-ionized water. Mix and fill up to 1 litre with de-ionized water.

#### 4.5 Standard solution chloride

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sodium chloride	NaCl	58.44	7647-14-5	164.84 g

#### **Preparation**

#### 100000 mg/L Chloride stock solution

Prepare a stock solution of 100000 mg/L Chloride: Dissolve accurately 164.84 g sodium chloride in 700 mL de-ionized water using a volumetric flask of 1000 mL. Fill up to 1 litre with de-ionized water.

#### **Chloride standard solution (dispenser)**

Prepare a standard solution according to the following table: take accurately x mL of the 100000 mg/L Cl stock solution and transfer into a volumetric flask of 1000 mL. Add deionized water up to the mark grade.

	Measuring range	Concentration REF2	Amount of stock solution to add to 1 litre
В	250 mg/L CI	2500 mg/L CI	25 mL
С	500 mg/L CI	5000 mg/L CI	50 mL
0	1000 mg/L CI	10000 mg/L CI	100 mL
V	5000 mg/L CI	10000 mg/L CI	100 mL
W	10000 mg/L CI	10000 mg/L CI	100 mL

### 4.6 Calibration solution

Blank solution - REF1

Use de-ionized water.

#### 4.7 Validation solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sodium chloride	NaCl	58.44	7647-14-5	164.84 g

#### **Preparation**

#### 100000 mg/L Chloride stock solution

Prepare a stock solution of 100000 mg/L Chloride: Dissolve accurately 164.84 g sodium chloride in 700 mL de-ionized water using a volumetric flask of 1000 mL. Fill up to 1 litre with de-ionized water.

#### Chloride validation solution

Prepare a standard solution according to the following table: take accurately x mL of the 100000 mg/L Cl stock solution and transfer into a volumetric flask of 1000 mL. Add deionized water up to the mark grade.

	Measuring range	Concentration Standard	Amount of stock solution to add to 1 litre
В	250 mg/L Cl	250 mg/L CI	2.5 mL
С	500 mg/L CI	500 mg/L CI	5.0 mL
0	1000 mg/L CI	1000 mg/L CI	10.0 mL
V	5000 mg/L CI	5000 mg/L CI	50.0 mL
W	10000 mg/L CI	10000 mg/L CI	100 mL

## 4.8 Cleaning solution (facultative)

The cleaning procedure should prevent any build-up of chemicals in the analyser. To obtain an effective cleaning procedure one has to test the cleaning solution and the cleaning interval for each application. Perform the selected cleaning solution and interval for a trial period, check then the effectiveness of the procedure and change if necessary.

	Change Information
Date: 14/12/2021	Previous version: Edition 2 to Edition 1.01
	Reason for Change
	ater consumption formation reagents
	Description of Change
	stimated consumption of water for rinse and dilution (chapter 4.2) ctra information regarding storage and quality of reagents (chapter 4.3)