

# **EZ2001 Total Chromium Analyser**

Method and reagent sheets

11/2021, Edition 1.01

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## 1. Legal information

Manufacturer: AppliTek NV/SA

Distributor: Hach Lange GmbH

The translation of the manual is approved by the manufacturer.

## 2. Analytical specifications

Please refer also to the respective technical datasheet at Hach Support Online.

Total Chromium - All specifications						
Analysis method	Colo	Colorimetric measurement at 546 nm using diphenylcarbazide method				
Parameter	Tota	ıl Cr				
	Star	ndard measurement cycle time: 20	minutes			
Cycle time	Inte	rnal dilution: + 5 min.				
	Exte	ernal dilution: + 5 – 10 min.				
Limit of detection (LOD)	ړ 2 ≥	ug/L				
Precision	Bett	er than 2% full scale range for sta	ndard test solutions			
Cleaning	Auto	omatic; frequency freely programm	nable			
Calibration	Auto	omatic, 2-point; frequency freely pr	rogrammable			
Validation	Auto	Automatic; frequency freely programmable				
	Iron (III) [(Fe)3+)], mercury [(Hg)2+] > 200 mg/L, molybdenum			$[(Mo)^{2+}] > 200 \text{ mg/L},$		
Interferences	vana	vanadium (V) [(V) <sup>5+</sup> ] > 5 mg/L. Large amounts of color and turbidity interfere. Fats, oil,				
	prot	proteins, surfactants and tar.				
Measuring ranges	% o	f range - Dilution	Low range (µg/L)	High range (µg/L)		
	Α	10% of standard range	2	50		
	В	25% of standard range	4	125		
	С	50% of standard range	5	250		
	0	standard range	10	500		
	1	internal MP dilution (factor 4)	160	2000		
	3	internal MP dilution (factor 10)	320	5000		
	4	internal MP dilution (factor 20)	1000	10000		

### 3. Analysis method

#### Summary

The determination of the chromium concentration in water is based on the reaction of hexavalent chromium with 1,5-diphenycarbazide in an acidic medium to form an intense coloured red-violet complex. The absorption is measured at 546 nm. Prior to the total chromium analysis, the sample is digested by use of a persulfate and an acid solution.

#### **Analysis steps**

The sample is mixed with persulfate and acid solution and heated to 120 °C (or up to 150 °C – programmable) in an oven during several minutes (standard 10 minutes; programmable up to 60 minutes). After digestion, the sample is cooled and transferred into the analysis vessel. Next, buffer solution is added to the sample and the initial absorbance value is measured. Next, the colour solution is added and the final absorbance value is determined. With the obtained absorbance values, the total chromium concentration can be calculated according to Beer's Law.

#### Calibration

The calibration procedure measures a REF1 Cr solution (channel 9, REF1 valve) and a REF2 Cr solution (channel 10, REF2 valve) to adapt the slope and offset factors by means of a two point calibration.

The calibration is performed in the MAIN method.

#### Remark

The methods cannot be started at the same time.

### 4. Reagents

## **A** CAUTION



Chemical exposure hazard. Obey laboratory safety procedures and wear all of the personal protective equipment appropriate to the chemicals that are handled. Read the safety data sheet from the supplier before bottles are filled or reagents are prepared. For laboratory use only. Make the hazard information known in accordance with the local regulations of the user.

## **A** CAUTION



Chemical exposure hazard. Dispose of chemicals and wastes in accordance with local, regional and national regulations.

### 4.1 Reagent overview and consumption

In the tables below, the products that are needed to prepare the reagents are listed. The product name, the formula, the molecular weight, the CAS No. and the amount needed to prepare 1 liter of the reagents is given. Check the consumption of the reagents (28 days) to adapt the volumes needed.

Product	Consumption	Consumption/28 days A rata 1 analysis/20 min	Recommended containers
Acid solution	~ 0.85 mL / analysis	~ 1.7 L	Plastic – 2.5 L
Persulfate solution	~ 0.25 mL / analysis	~ 0.5 L	Plastic – 2.5 L
Buffer solution	~ 0.85 mL / analysis	~ 1.7 L	Plastic – 2.5 L
Colour solution	~ 2.0 mL / analysis	~ 4.0 L	Glass-Amber – 2.5 L
REF1 solution	~ 0.5 L / calibration	1	Plastic – 1 L
REF2 solution	~ 0.5 L / calibration	1	Plastic – 1 L

### 4.2 DI-water overview and consumption

	Rinse water (mL/analysis) Type I	Dilution water (mL/analysis) Type I	Total (mL/analysis)	Consumption/28 days A rata 1 analysis/20 min
Α	N.A.	N.A.	N.A.	N.A.
В	N.A.	N.A.	N.A.	N.A.
С	N.A.	N.A.	N.A.	N.A.
0	N.A.	N.A.	N.A.	N.A.
1	55 mL	20 mL	75 mL	152 L
3	55 mL	20 mL	75 mL	152 L
4	55 mL	20 mL	75 mL	152 L

#### Remark

The indicated volumes are an estimation of the consumption for rinse and dilution water, based on a standard operating procedure, as defined in the specifications of the EZ analyser. Please be aware that, depending on the sample matrix, the rinse water volumes might increase.

### 4.3 Storage and quality of chemicals

#### **Quality of chemicals**

All chemicals should be of Reagent grade, ACS grade or better (\*). The use of pro analysis chemicals is recommended. Poor quality of the reagents can affect the analyser performance.

(\*) Analytical Reagent (AR), Guaranteed Reagent (GR), UNIVAR, AnalaR, Premium Reagent (PR), ReagentCertified ACS reagent, ACS Plus reagent, puriss p.a. ACS reagent, ReagentPlus®, TraceCERT®, Suprapur®, Ultrapur®, or better are also possible.

### **Quality of DI-water**

All EZ analysers are tested with standard solutions, reagents and dilution water prepared using type I water or better as defined by ASTM D1193-91.

To achieve the specifications as stated on the data sheet, method and reagents sheet and acceptance test reports, the same water quality (or better) must be used for the preparation of the standard solutions, reagents and dilution water.

Additionally the water used for the preparation of the standard solutions for an EZ analyser must be free of the parameter or any of the interferences for the method of that EZ analyser.

### **Storage of Reagents**

While operating the instrument, keep in mind the reagent requirements as stated in the reagent overview, the chapters below and/or in the data sheet of the instrument.

## **ACAUTION**



For longer-term storage: Store the reagents cold; Store the reagents in the dark;

If applicable: Store the reagents in a fridge during operation

### **ACAUTION**



Refresh the reagents after one month (unless stated differently in the chapters below).

Do not mix old reagents with freshly prepared reagents. Remove old reagents from the container before adding freshly prepared reagents.

### 4.4 Acid solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sulfuric acid 96%	H <sub>2</sub> SO <sub>4</sub>	98.08	7664-93-9	56 mL

### Preparation

Dilute 56 mL sulfuric acid ( $H_2SO_4$ , 96%) in 500 mL de-ionized water. Mix and fill up to 1 litre with de-ionized water.

### 4.5 Persulfate solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sodium persulfate	Na <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	238.09	7775-27-1	10 g

### **Preparation**

Add 10 g of sodium persulfate ( $Na_2S_2O_8$ ) in 500 mL de-ionized water. Mix and fill up to 1 litre with de-ionized water.

### 4.6 Buffer solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sulfuric acid 96%	H <sub>2</sub> SO <sub>4</sub>	98.08	7664-93-9	56 mL

### **Preparation**

Dilute 56 mL sulfuric acid ( $H_2SO_4$ , 96%) in 500 ml de-ionized water. Mix and fill up to 1 litre with de-ionized water.

### 4.7 Colour solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
1,5-Diphenylcarbazide	C <sub>13</sub> H <sub>14</sub> N <sub>4</sub> O	242.28	140-22-7	1 g
Acetone (analytic)	C <sub>3</sub> H <sub>6</sub> O	58.08	67-64-1	500 mL
Acetic acid (glacial)	C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	60.05	64-19-7	2 mL

### **Preparation**

Dissolve 1 g of 1.5-Diphenylcarbazide ( $C_{13}H_{14}N_4O$ ) completely in 500 mL acetone ( $C_3H_6O$ ). Add carefully 2 mL acetic acid ( $C_2H_4O_2$ ) and fill up to 1 litre with de-ionized water. The pH value of this solution should be below 4.

### 4.8 Calibration solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Potassium dichromate	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	294.18	7778-50-9	1.4143 g

### **Preparation**

### 500 mg/L Cr stock solution

Prepare a stock solution of 500 mg/L Cr: Dissolve accurately 1.4143 g potassium dichromate ( $K_2Cr_2O_7$ ) in 300 mL de-ionized water using a volumetric flask of 1000 mL. Fill up to 1 litre with de-ionized water.

### Cr standard solution - REF2

Prepare a standard solution for calibration according to the following table: take accurately x mL of the 500 mg/L Cr stock solution and transfer into a plastic volumetric flask of 1 litre. Add de-ionized water up to the mark grade.

	Measuring range	Concentration REF2	Amount of stock solution to add to 1 litre
Α	50 μg/L Cr	50 μg/L Cr	0.10 mL
В	125 μg/L Cr	125 μg/L Cr	0.25 mL
С	250 μg/L Cr	250 μg/L Cr	0.50 mL
0	500 μg/L Cr	500 μg/L Cr	1.0 mL
1	2000 μg/L Cr	2000 μg/L Cr	4.0 mL
3	5000 μg/L Cr	5000 μg/L Cr	10.0 mL
4	10000 μg/L Cr	10000 μg/L Cr	20.0 mL

### Cr standard solution - REF1

Prepare a standard solution of 0 µg/L Cr. Use de-ionized water.

### 4.9 Cleaning solution (facultative)

The cleaning procedure should prevent any build-up of chemicals in the analyser. To obtain an effective cleaning procedure one has to test the cleaning solution and the cleaning interval for each application. Perform the selected cleaning solution and interval for a trial period, check then the effectiveness of the procedure and change if necessary.

	Change Information
Date: 02/11/2021	Previous version: Edition 5 to Edition 1.01
	Reason for Change
	ater consumption formation reagents
	Description of Change
	stimated consumption of water for rinse and dilution (chapter 4.2) ktra information regarding storage and quality of reagents (chapter 4.3)