

EZ3007 Fluoride Analyser

Method and reagent sheets

01/2023, Edition 1.01

1.	Legal information	3
2.	Analytical specifications	3
3.	Analysis method	4
4.	Reagents	5
		_
4.1	Reagent overview and consumption	5
4.2	DI-water overview and consumption	5
12	Storage and quality of chemicals	6
4.4	Buffer solution	7
4 E	Calibration solution	0
4.5		ŏ
4.6	Cleaning solution (facultative)	8

1. Legal information

Manufacturer: AppliTek NV/SA

Distributor: Hach Lange GmbH

The translation of the manual is approved by the manufacturer.

2. Analytical specifications

Please refer also to the respective technical datasheet at Hach Support Online.

Fluoride - All specifications				
Analysis method	Disco	ntinuous, direct measurement	by combined ion-selective	electrode
Parameter	Fluori	de		
Cycle time	Stand	ard measurement cycle time: 5	5 minutes	
Limit of detection (LOD)	≤ 0.05	i mg/L		
Precision/Repeatability	Better than 2% full scale range for standard test solutions			
Cleaning	Autom	natic; frequency freely program	mable	
Calibration	Automatic, 2-point; frequency freely programmable			
Validation	Automatic; frequency freely programmable			
Interferences	rences Metal ions like aluminium $[(AI)^{3+}] > 72 \text{ mg/L}$, calcium $[(Ca)^{2+}] > 108 \text{ mg/L}$ and i $[(Fe)^{2+}/(Fe)^{3+}] > 150 \text{ mg/L}$. Fats, oil, proteins, surfactants and tar.			-
Measuring ranges	% of r	ange - Dilution	Low range (mg/L)	High range (mg/L)
	B 25% of standard range		0.05	2.5
	С	50% of standard range	0.1	5
	0 standard range 0.2 10			10

3. Analysis method

Summary

The Fluoride (F⁻) concentration is determined by a discontinues, direct measurement using an ionselective electrode.

Analysis steps

The analysis vessel is flushed with fresh sample. The TISAB [T(otal) I(onic) S(trength) A(djustment) B(uffer)] solution is added to the sample in order to adjust the pH and to assure the total strength of the sample. The potential of the sample solution is measured. With the obtained value, the analyzer calculates the fluoride concentration in the sample.

Calibration

The calibration procedure measures a REF1 F solution (channel 9, REF1 valve) and a REF2 F solution (channel 10, REF2 valve) to adapt the slope and offset factors by means of a two point calibration.

The calibration is performed in the MAIN method.

Remark

The methods cannot be started at the same time.

4. Reagents

ACAUTION

Chemical exposure hazard. Obey laboratory safety procedures and wear all of the personal protective equipment appropriate to the chemicals that are handled. Read the safety data sheet from the supplier before bottles are filled or reagents are prepared. For laboratory use only. Make the hazard information known in accordance with the local regulations of the user.



ACAUTION

Chemical exposure hazard. Dispose of chemicals and wastes in accordance with local, regional and national regulations.

4.1 Reagent overview and consumption

In the tables below, the products that are needed to prepare the reagents are listed. The product name, the formula, the molecular weight, the CAS No. and the amount needed to prepare 1 litre of the reagents is given. Check the consumption of the reagents (28 days) to adapt the volumes needed.

Product	Consumption	Consumption/28 days A rata 1 analysis/5 min	Recommended containers
Buffer solution	~ 0.1 mL / analysis	< 1 L	Plastic – 2.5 L
REF1 solution	~ 0.5 L / calibration	/	Plastic – 1 L
REF2 solution	~ 0.5 L / calibration	/	Plastic – 1 L

4.2 DI-water overview and consumption

	Rinse water (mL/analysis) Type I	Dilution water (mL/analysis) Type I	Total (mL/analysis)	Consumption/28 days A rata 1 analysis / 5 min
В	N.A.	N.A.	N.A.	N.A.
С	N.A.	N.A.	N.A.	N.A.
0	N.A.	N.A.	N.A.	N.A.

4.3 Storage and quality of chemicals

Quality of chemicals

All chemicals should be of Reagent grade, ACS grade or better (*). The use of pro analysis chemicals is recommended. Poor quality of the reagents can affect the analyser performance.

(*) Analytical Reagent (AR), Guaranteed Reagent (GR), UNIVAR, AnalaR, Premium Reagent (PR), ReagentCertified ACS reagent, ACS Plus reagent, puriss p.a. ACS reagent, ReagentPlus[®], TraceCERT[®], Suprapur[®], Ultrapur[®], or better are also possible.

Quality of DI-water

All EZ analysers are tested with standard solutions, reagents and dilution water prepared using type I water or better as defined by ASTM D1193-91.

To achieve the specifications as stated on the data sheet, method and reagents sheet and acceptance test reports, the same water quality (or better) must be used for the preparation of the standard solutions, reagents and dilution water.

Additionally the water used for the preparation of the standard solutions for an EZ analyser must be free of the parameter or any of the interferences for the method of that EZ analyser.

Storage of Reagents

While operating the instrument, keep in mind the reagent requirements as stated in the reagent overview, the chapters below and/or in the data sheet of the instrument.



For longer-term storage: Store the reagents cold; Store the reagents in the dark;

If applicable: Store the reagents in a fridge during operation



Refresh the reagents after one month (unless stated differently in the chapters below).

Do not mix old reagents with freshly prepared reagents. Remove old reagents from the container before adding freshly prepared reagents.

4.4 Buffer solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Glacial acetic acid	$C_2H_4O_2$	60.05	64-19-7	58 mL
Sodium Hydroxide	NaOH	40.00	1310-73-2	37 g
Sodium Chloride	NaCl	58.44	7647-14-5	58.4 g
DCTA	C ₁₄ H ₂₂ N ₂ O ₈ * H ₂ O	365.35	125572-95-4	5.0 g

2M NaOH solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sodium Hydroxide	NaOH	40.00	1310-73-2	80 g

Preparation

Add carefully 58 mL glacial acetic acid ($C_2H_4O_2$ 100%) to 600 mL de-ionized water and dissolve. Add 37 g sodium hydroxide (NaOH) and after dissolving let the solution cool down. Add 58.4 g sodium chloride (NaCl) and 5.0 g DCTA (1,2 Diaminocyclohexane-N,N,N',N'-tetraacetic acid monohydrate) and dissolve.

The pH value of this solution should be 5.5 ± 0.1 . If this is not the case, adjust the pH to 5.5 ± 0.1 with the sodium hydroxide (NaOH 2M) solution. The amount of 2M NaOH that needs to be added will not be exceeding 25 mL. Fill up to 1 litre with de-ionized water.

4.5 Calibration solution

Products	Formula	MW (g/mol)	CAS No.	1 litre solution
Sodium fluoride	NaF	41.98	7681-49-4	2.2095 g

Preparation

1000 mg/L Fluoride stock solution

Prepare a stock solution of 1000 mg/L Fluoride: Dissolve accurately 2.2095 g sodium fluoride in 300 mL de-ionized water using a volumetric flask of 1000 mL. Fill up to 1 litre with de-ionized water.

F standard solution – REF2

Prepare a standard solution for calibration according to the following table: take accurately x mL of the 1000 mg/L F stock solution and transfer into a volumetric flask of 1000 mL. Add de-ionized water up to the mark grade.

	Measuring range	Concentration REF2	Amount of stock solution to add to 1 litre
В	2.5 mg/L F	2.5 mg/L F	2.5 mL
С	5 mg/L F	5 mg/L F	5.0 mL
0	10 mg/L F	10 mg/L F	10.0 mL

F standard solution – REF1

Prepare a standard solution for calibration according to the following table: take accurately x mL of the 1000 mg/L F stock solution and transfer into a volumetric flask of 1000 mL. Add de-ionized water up to the mark grade.

	Measuring range	Concentration REF1	Amount of stock solution to add to 1 litre
В	2.5 mg/L F	0.25 mg/L F	0.25 mL
С	5 mg/L F	0.5 mg/L F	0.5 mL
0	10 mg/L F	1 mg/L F	1 mL

4.6 Cleaning solution (facultative)

The cleaning procedure should prevent any build-up of chemicals in the analyser. To obtain an effective cleaning procedure one has to test the cleaning solution and the cleaning interval for each application. Perform the selected cleaning solution and interval for a trial period, check then the effectiveness of the procedure and change if necessary.

	Change Information
Date: 09/01/2023	Previous version: Edition 4 to Edition 1.01
	Reason for Change
- Addition of inf	ater consumption formation reagents eparation method of buffer
	Description of Change
 Addition of ex 	atimated consumption of water for rinse and dilution (chapter 4.2) (tra information regarding storage and quality of reagents (chapter 4.3) (eparation method Buffer solution, due to current complex preparation method